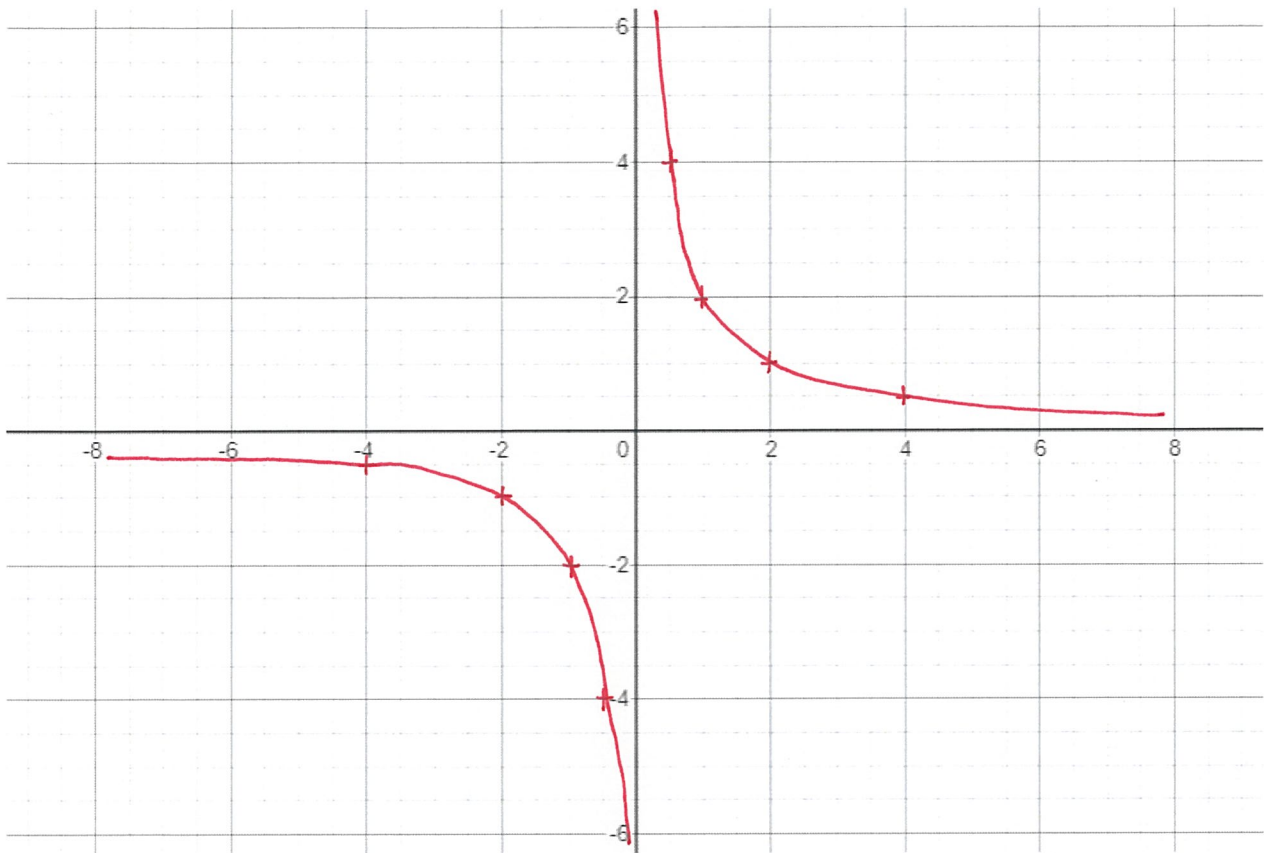
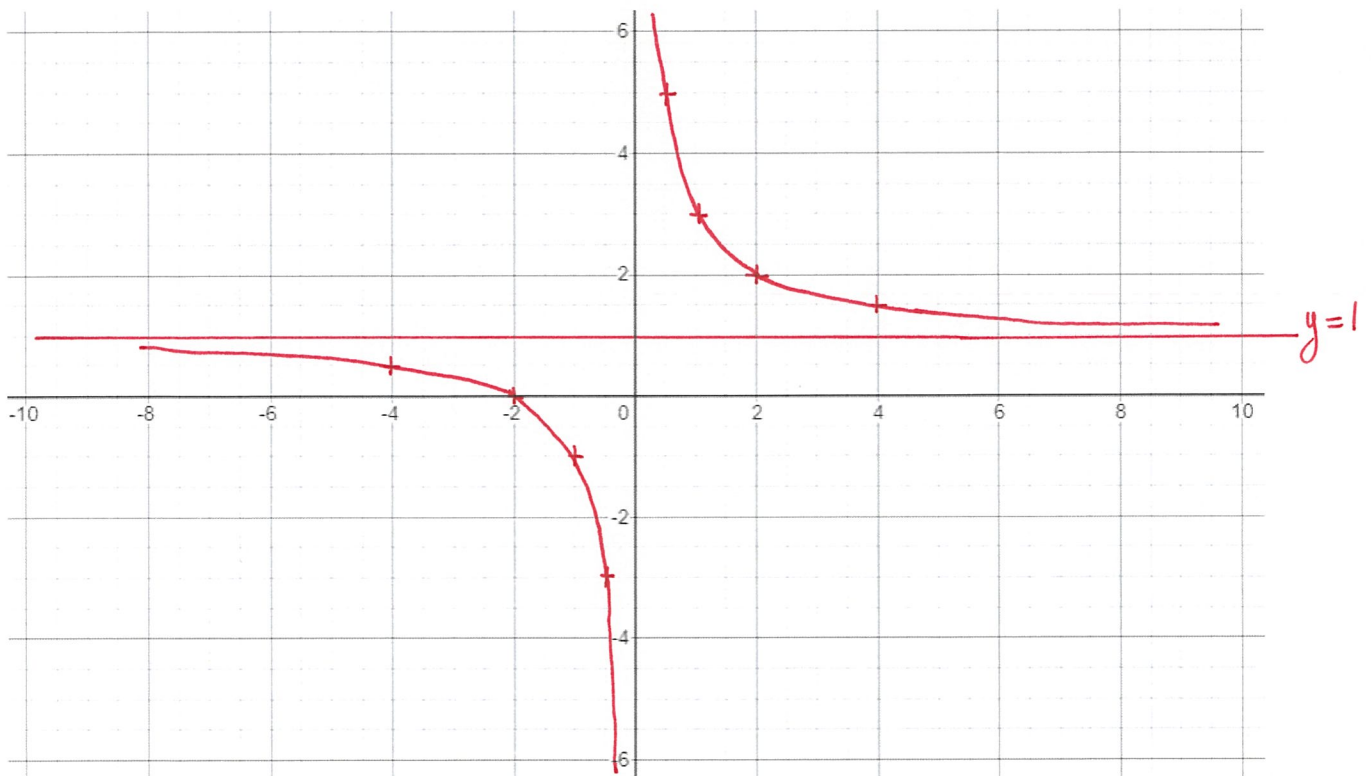


## THE EQUATION $y = k/x$ AND INVERSE VARIATION

2 Draw the graph of  $y = \frac{2}{x}$ . Write the equations of its asymptotes.

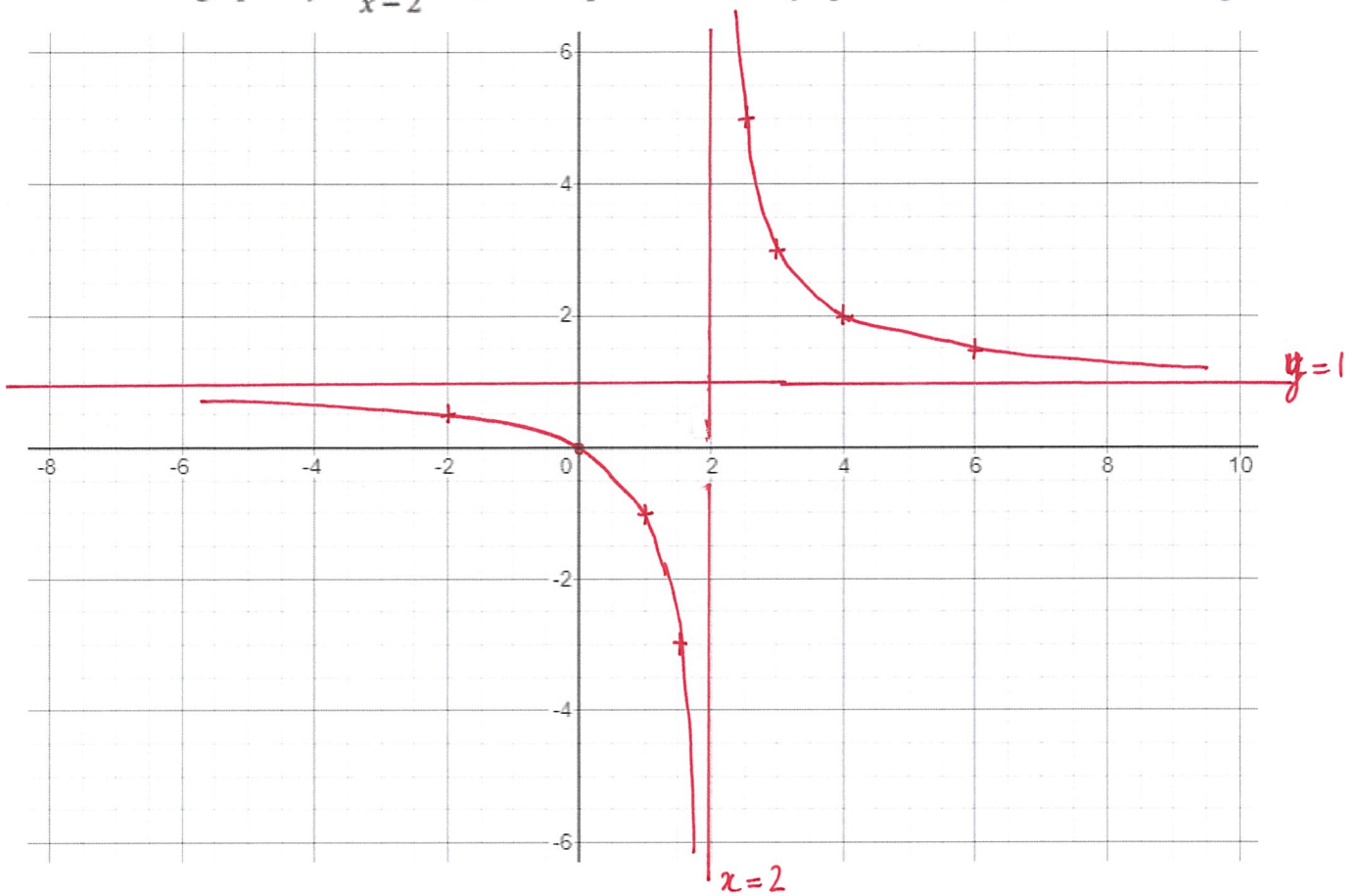


3 Draw the graph of  $y = \frac{x+2}{x}$ . Write the equations of its asymptotes, the domain and the range.



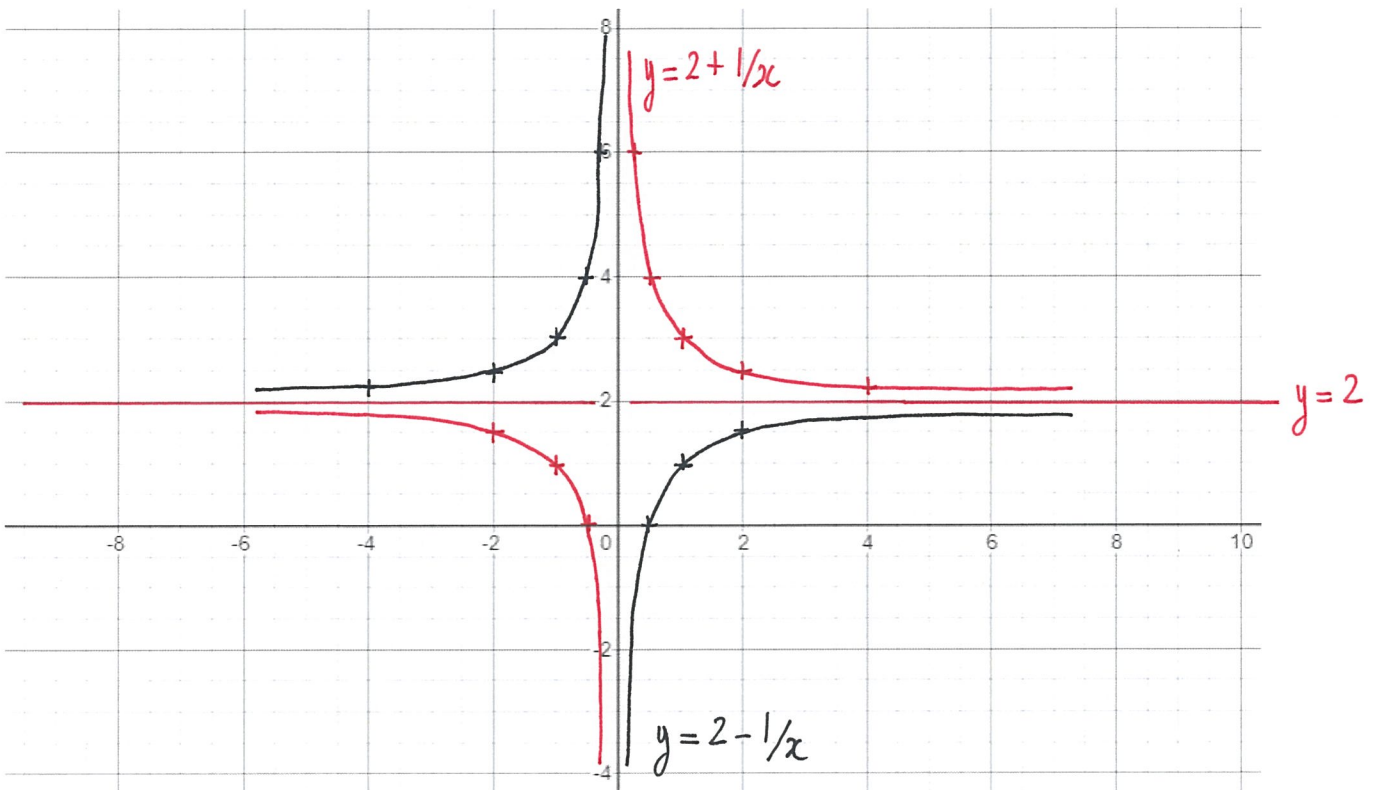
## THE EQUATION $y = k/x$ AND INVERSE VARIATION

4 Draw the graph of  $y = \frac{x}{x-2}$ . Write the equations of its asymptotes, the domain and the range.



## THE EQUATION $y = k/x$ AND INVERSE VARIATION

- 5 (a) On the same set of axes, draw the graphs of  $y = 2 + \frac{1}{x}$  and  $y = 2 - \frac{1}{x}$ .  
(b) Do these graphs ever intersect?  
(c) Comment on their asymptotes.



b) The graphs do not intersect.

c) Their asymptotes are the same.

## THE EQUATION $y = k/x$ AND INVERSE VARIATION

- 6 (a) In an experiment it is found that at a temperature of  $100^{\circ}\text{C}$ , 2 litres of argon gas is at a pressure of 15.28 atmospheres. If this gas obeys Boyle's law,  $PV = k$ , where  $V$  is in litres and  $P$  is in atmospheres, then find the value of  $k$ .
- (b) If the volume was expanded to 4 litres with the temperature held at  $100^{\circ}\text{C}$ , then what would be the expected pressure?
- (c) If the pressure was increased to 90 atmospheres with the temperature held at  $100^{\circ}\text{C}$ , then what would be the expected volume?

$$\text{a) } PV = k \quad \text{so} \quad k = 15.28 \times 2 = 30.56$$

$$\text{b) } P = \frac{k}{V} = \frac{30.56}{4} = 7.64 \text{ atm}$$

$$\text{c) } V = \frac{k}{P} = \frac{30.56}{90} \approx 0.34 \text{ litre}$$